This Page Is Inserted by IFW Operations and is not a part of the Official Record

BEST AVAILABLE IMAGES

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

IMAGES ARE BEST AVAILABLE COPY.

As rescanning documents will not correct images, please do not report the images to the Image Problem Mailbox.

THE BRITISH LIBRARY SCIENCE REFERENCE AND INFORMATION SERVICE



WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 6: WO 95/35265 (11) International Publication Number: C03C 13/00, 13/06 **A1** 28 December 1995 (28.12.95) (43) International Publication Date:

(81) Designated States: AU, BR, CA, CN, CZ, FI, HU, JP, KR, PCT/EP95/02375 (21) International Application Number: MX, NO, NZ, PL, SI, SK, US, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, 19 June 1995 (19.06.95) (22) International Filing Date:

19 June 1994 (19.06.94) P 44 21 120.1 DE DE 195 03 170.9 1 February 1995 (01.02.95)

(71) Applicant (for all designated States except US): ISOVER SAINT-GOBAIN [FR/FR]; Les Miroirs, 18, avenue

d'Alsace, F-92400 Courbevoie (FR).

(72) Inventors; and (75) Inventors/Applicants (for US only): BERNARD, Jean, Luc [FR/FR]; 51, rue André-Oudin, Giencourt, F-60600 Clermont (FR). DE MERINGO, Alain [FR/FR]; 294, rue Saint-Jacques, F-75005 Paris (FR). ROUYER, Elisabeth [FR/FR]; 32 bis, rue de l'Alma, F-92600 Asnières (FR). FURTAK, Hans [DE/DE]; Im Oberkämmerer 35, D-67346 Speyer am Rhein (DE).

(74) Agent: KADOR & PARTNER; Comeliusstrasse 15, D-80469 Munich (DE).

Published

With international search report.

Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.

(54) Title: MINERAL-FIBER COMPOSITIONS

(57) Abstract

(30) Priority Data:

A biologically degradable mineral-fiber composition characterized by the following constituents in percent by weight: SiO2: 45 to 60, Al₂O₃: less than 2, CaO 7: to 18, MgO: 4 to 10, Na₂O: 7 to 20, K₂O: 0 to 4, B₂O₃: 1 to 12, P₂O₅: 0 to 4, diverse: 0 to 5, Na₂O: + K₂O: 7 to 24, CaO + MgO: more than 15.5 and up to 25, BaO: 0 to 5, TiO₂: 0 to 4, Cr₂O₃: 0 to 1.5, Fe₂O₃: 0 to 3.

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
AU	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Kenya	RO	Romania
CA	Canada	KG	Kyrgystan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic	SD	Sudan
CG	Congo		of Korea	SE	Sweden
CH	Switzerland	KR	Republic of Korea	SI	Slovenia
CI	Côte d'Ivoire	KZ	Kazakhstan	SK	Slovakia
CM	Cameroon	LI	Liechtenstein	SN	Senegal
CN	China	LK	Sri Lanka	TD	Chad
CS	Czechoslovakia	LU	Luxembourg	TG	Togo
CZ	Czech Republic	LV	Latvia	TJ	Tajikistan
DE	Germany	MC	Monaco	TT	Trinidad and Tobago
DK	Denmark	MD	Republic of Moldova	UA	Ukraine
ES	Spain	MG	Madagascar	US	United States of America
FI	Finland	ML	Mali	UZ	Uzbekistan
FR	France	MN	Mongolia	VN	Viet Nam
GA	Gabon	••••			

WO 95/35265

Mineral-fiber compositions

The present invention relates to a mineral-fiber composition that is biologically degradable.

The prior art describes some mineral-fiber compositions which are said to be biologically degradable.

The biological degradability of mineral-fiber compositions is of great importance because various studies point out that mineral fibers with very small diameters in the range of less than 3 microns can be carcinogenic, while biologically degradable mineral fibers of such dimensions show no carcinogenicity.

However not only the biological degradability is of crucial importance but also the mechanical and thermal properties of the mineral fibers, or the products produced therefrom, the resistance of the mineral fibers and the processibility of the mineral-fiber composition.

For example mineral fibers are used to a great extent for insulation purposes. For these applications sufficient moisture-resistance is necessary.

Also, the mineral-fiber composition must permit processibility by known methods for producing mineral fibers with a small diameter, for example the centrifugal technique, in particular the inner centrifugal technique (this technique is described for example in US-PS 4 203 745).

The invention is based on the problem of providing a novel mineral-fiber composition that is characterized by

biological degradability, has good stability or resistance to moisture and is easy to process.

The invention is based on the finding that this problem can be solved by a mineral-fiber composition that has considerable amounts of alkali oxides and alkaline-earth oxides, and optionally phosphorus oxide.

It has turned out that such a mineral-fiber compositi- on fulfills the combination of the necessary properties, namely biological degradability, resistance to moisture and good processibility.

The object of the invention is a mineral-fiber composition that is biologically degradable, characterized by the following constituents in percent by weight:

SiO ₂				45		to	60
Al ₂ O ₃				le	SS	tha	n 2
CaO				7		to	18
MgO				4		to	10
Na ₂ O				7		to	20
K ₂ O				0		to	4
B ₂ O ₃				1		to	12
P ₂ O ₅				0		to	4
Diverse				0		to	5
$Na_2O + K_2O$				7		to	24
CaO + MgO	more	than	15.5	and	up	to	25
BaO				0		to	5
TiO ₂				0		to	4
Cr ₂ O ₃				0	to) 1	5
Fe ₂ O ₃				0	t	:0	3.

The inventive mineral-fiber compositions are processible by the centrifugal technique. The obtained fibers have good resistance to moisture. Surprisingly enough, the mineral-fiber compositions show biological degradability. The mean fiber diameter is preferably 10 microns or less, and is in particular between 2.5 and 5 microns.

The inventive mineral-fiber compositions preferably have the following constituents in percent by weight:

SiO ₂	50	to	58
Al ₂ O ₃	less	thar	n 2
CaO	10	to	18
MgO	4	to	8
Na ₂ O	10	to	18
K ₂ O	0	to	2
B ₂ O ₃	3	to	12
P ₂ O ₅	0.5	to	4
Diverse	0	to	2
$Na_2O + K_2O$	10	to	21
CaO + MgO	16	to	24
BaO	0	to	4
TiO ₂	0	to	3
Cr ₂ O ₃	0	to	1
Fe ₂ O ₃	0	to	2.

The inventive mineral-fiber compositions have in particular the following constituents in percent by weight:

SiO ₂	50	to 57
Al ₂ O ₃	0.5	to 1.5
CaO	11	to 16
MgO	4.5	to 6
Na ₂ O	12	to 17

PCT/EP95/02375

K ₂ O	0.5	to 1
B ₂ O ₃	5	to 11
P ₂ O ₅	1	to 3
Diverse	0.5	to 1.0
$Na_2O + K_2O$	11	to 17
CaO + MgO	16	to 22
BaO	0	to 3
TiO ₂	0	to 2
Cr ₂ O ₃	0	to 0.5
Fe ₂ O ₃	0	to 1.5.

The inventive mineral-fiber compositions preferably have less than 55% silicon dioxide.

It is also particularly preferred that the compositions contain more than 5 percent by weight, in particular more than 6 percent by weight, magnesium oxide.

Barium oxide is preferably added in exchange for calcium oxide.

Biological degradability can be increased by adding phosphorus pentoxide. The inventive compositions therefore preferably contain at least 0.5 percent by weight P_2O_5 .

It is advantageous to add titanium oxide, chromium oxide and/or iron oxide to reduce the corrosive properties of the melt.

The moisture-resistance of the inventive mineral-fiber compositions was determined by a standard method known as the DGG method. In the DGG method 10 g finely ground mineral with a grain size between about 360 and 400 microns is held at the boiling point for five hours in 100 ml water. After quick cooling of the material the solution is filte-

red and a certain volume of the filtrate evaporated to dryness. The weight of the thus obtained dry material permits the amount of mineral dissolved in the water to be calculated. The amount is stated in milligrams per gram of tested mineral.

The biological degradability of the inventive mineral compositions was tested by introducing 1 g of the mineral powder, as described for the DGG method, into a physiological solution with the composition stated below and a pH value of 7.4:

NaCl	6.78
NH₄Cl	0.535
NaHCO ₃	2.268
NaH ₂ PO ₄ H ₂ O	0.166
(Na ₃ citrate) 2H ₂ O	0.059
Glycine	0.450
H ₂ SO ₄	0.049
CaCl ₂	0.022

Dynamic test conditions were selected as are described in Scholze and Conradt. The flow rate was 300 ml/day. The duration of the test was 14 days. The results are stated as percent of SiO_2 in the solution x 100 after 14 days.

The invention shall be described in more detail in the following with reference to examples.

Example 1

A composition was produced with the following constituents in percent by weight:

SiO ₂	56
Al ₂ O ₃	0.5
CaO	15
MgO	4.0
Na ₂ O	16.2
K ₂ O	0.8
B_2O_3	5.5
P_2O_5	1.5
Diverse	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 32 mg/g was determined.

The above-described test for biological degradability yielded a value of 615.

Example 2

A composition was produced with the following constituents in percent by weight:

SiO ₂	54.5
Al ₂ O ₃	0.5
CaO	15
MgO	4.0
Na ₂ O	16.2

K ₂ O	0.8
B ₂ O ₃	5.5
P ₂ O ₅	3.0
Diverse	0.5.

Using the above-described DGG method a value of 32 mg/g was determined.

The above-described test for biological degradability yielded a value of 690.

Example 3

A composition was produced with the following constituents in percent by weight:

SiO ₂	56
Al ₂ O ₃	0.5
CaO	13
MgO	6
Na ₂ O	16.2
K ₂ O	0.8
B ₂ O ₃	5.5
P ₂ O ₅	1.5
Diverse	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 32 mg/g was determined.

The above-described test for biological degradability yielded a value of 615.

Example 4

A composition was produced with the following constituents in percent by weight:

SiO ₂	54.5
Al ₂ O ₃	. 0.5
CaO	13
MgO	6
Na ₂ O	16.2
K ₂ O	0.8
B_2O_3	5.5
P ₂ O ₅	3
Diverse ·	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 32 mg/g was determined.

The above-described test for biological degradability yielded a value of 690.

Example 5

A composition was produced with the following constituents in percent by weight:

56

Al_2O_3	0.5
CaO	16
MgO	6
Na ₂ O	13.2
K ₂ O	0.8
B_2O_3	5.5
P ₂ O ₅	1.5
Diverse	0.5.

Using the above-described DGG method a value of 22 mg/g was determined.

The above-described test for biological degradability yielded a value of 585.

Example 6

A composition was produced with the following constituents in percent by weight:

SiO ₂	54.5
Al ₂ O ₃	0.5
CaO	16
MgO	6
Na ₂ O	13.2
K ₂ O	0.8
B ₂ O ₃	5.5
P ₂ O ₅	3
Diverse	0.5.

Using the above-described DGG method a value of 22 mg/g was determined.

The above-described test for biological degradability yielded a value of 660.

Example 7

A composition was produced with the following constituents in percent by weight:

SiO ₂	54.5
Al ₂ O ₃	1
CaO	16
MgO .	6
Na ₂ O	13.2
K ₂ O	0.8
B_2O_3	6.5
P ₂ O ₅	1.5
Diverse	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 17 mg/g was determined.

The above-described test for biological degradability yielded a value of 570.

Example 8

A composition was produced with the following constituents in percent by weight:

SiO ₂	53
Al ₂ O ₃	1
CaO	16
MgO	6
Na ₂ O	13.2
K ₂ O	0.8
B ₂ O ₃	6.5
P_2O_5	3
Diverse	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 17 mg/g was determined.

The above-described test for biological degradability yielded a value of 645.

Example 9

A composition was produced with the following constituents in percent by weight:

SiO ₂	50.5
Al ₂ O ₃	1.5
CaO	16
MgO	8
Na ₂ O	12.2

K ₂ O	0.8
B_2O_3	6.5
P_2O_5	4
Diverse	0.5

Using the above-described DGG method a value of 7 mg/g was determined.

The above-described test for biological degradability yielded a value of 660.

Example 10

A composition was produced with the following constituents in percent by weight:

SiO ₂	50.5
Al ₂ O ₃	1.5
CaO	18
MgO	6
Na ₂ O	10.7
K ₂ O	0.8
B_2O_3	8
P ₂ O ₅	4
Diverse	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 7 mg/g was determined.

The above-described test for biological degradability yielded a value of 660.

Example 11

A composition was produced with the following constituents in percent by weight:

SiO ₂	55
Al ₂ O ₃	1
CaO	11
MgO	5
Na ₂ O	14.2
K₂O	0.8
B ₂ O ₃	11.5
P ₂ O ₅	1
Diverse	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 31 mg/g was determined.

The above-described test for biological degradability yielded a value of 600.

Example 12

A composition was produced with the following constituents in percent by weight:

SiO₂

Al ₂ O ₃	0.5
CaO	11
MgO	5
Na ₂ O	14.2
K ₂ O	0.8
B ₂ O ₃	10.5
P ₂ O ₅	1
Diverse	0.5.

Using the above-described DGG method a value of 36 mg/g was determined.

The above-described test for biological degradability yielded a value of 620.

Example 13

A composition was produced with the following constituents in percent by weight:

SiO ₂	58.0
Al ₂ O ₃	0.5
CaO	13.0
MgO	7.0
$Na_2O + K_2O$	13.5
B ₂ O ₃	8.0.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 21 mg/g was determined.

The above-described test for biological degradability yielded a value of 515.

Example 14

A composition was produced with the following constituents in percent by weight:

SiO ₂	57.0
Al ₂ O ₃	0.5
Fe ₂ O ₃	1.0
CaO .	13.0
MgO	7.0
$Na_2O + K_2O$	13.5
B ₂ O ₃	8.0.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 22 mg/g was determined.

The above-described test for biological degradibilty yielded a value of 480.

Example 15

A composition was produced with the following constituents in percent by weight:

SiO ₂	58.0
Al ₂ O ₃	0.5
CaO	9.5
MgO	7.0
$Na_2O + K_2O$	17.0
B ₂ O ₃	8.0.

Using the above-described DGG method a value of 36 mg/g was determined.

The above-described test for biological degradability yielded a value of 550.

Example 16

A composition was produced with the following constituents in percent by weight:

SiO ₂	58.0
Al ₂ O ₃	0.5
CaO	13.0
MgO	7.0
$Na_2O + K_2O$	17.0
B ₂ O ₂	4.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 27 mg/g was determined.

The above-described test for biological degradability yielded a value of 515.

Example 17

A composition was produced with the following constituents in percent by weight:

SiO ₂	57.5
Al ₂ O ₃	0.5
CaO	13.5
MgO	6.5
$Na_2O + K_2O$	17.0
B ₂ O ₃	4.5
Cr ₂ O ₃	0.5.

This mineral composition could be processed by the centrifugal technique.

Using the above-described DGG method a value of 25 mg/g was determined.

The above-described test for biological degradability yielded a value of 490.

Claims

1. A mineral-fiber composition that is biologically degradable, characterized by the following constituents in percent by weight:

SiO ₂				45	to	60
Al ₂ O ₃				less	tha	an 2
CaO				7	to	18
MgO				4	to	10
Na ₂ O				7	to	20
K ₂ O				0	to	4
B ₂ O ₃				1	to	12
P ₂ O ₅				0	to	4
Diverse				0	to	5
$Na_2O + K_2O$				7	to	24
CaO + MgO	more	than	15.5	and up	p to	25
BaO				0	to	5
TiO ₂				0	to	4
Cr ₂ O ₃				0 t	.0	1.5
Fe ₂ O ₃				0	to	3.

2. The mineral-fiber composition of claim 1, characterized by the following constituents in percent by weight:

SiO ₂	50	to	58
Al ₂ O ₃	less	thar	1 2
CaO	10	to	18
MgO	4	to	8
Na ₂ O	10	to	18
K ₂ O	0	to	2

B_2O_3	3	to	12
P_2O_5	0.5	to	4
Diverse	0	to	2
$Na_2O + K_2O$	10	to	21
CaO + MgO	16	to	24
BaO	0	to	4
TiO ₂	0	to	3
Cr ₂ O ₃	0	to	1
Fe ₂ O ₃	0	to	2.

3. The mineral-fiber composition of claim 1 or 2, characterized by the following constituents in percent by weight:

SiO ₂	50	to 57
Al ₂ O ₃	0.5	to 1.5
CaO	11	to 16
MgO	4.5	to 6
Na ₂ O	12	to 17
K ₂ O -	0.5	to 1
B_2O_3	5	to 11
P ₂ O ₅	1	to 3
Diverse	0.5	to 1.0
$Na_2O + K_2O$	11	to 17
CaO + MgO	16	to 22
BaO	0	to 3
\mathtt{TiO}_2	0	to 2
Cr ₂ O ₃	0	to 0.5
Fe ₂ O ₃	0	to 1.5.

- 4. The mineral-fiber composition of any of claims 1 to 3, characterized in that the content of silicon dioxide is less than 55 percent by weight.
- 5. The mineral-fiber composition of any of claims 1 to 4, characterized in that the content of magnesium dioxide is more than 5 percent by weight.
- 6. The mineral-fiber composition of any of claims 1 to 5, characterized in that the content of magnesium dioxide is more than 6 percent by weight.

INTERNATIONAL SEARCH REPORT

PCT/EP 95/02375

A. CLASSIFICATION OF SUBJECT MATTER IPC 6 C03C13/00 C03C13/06

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols) IPC 6 CO3C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUM	ENTS CONSIDERED TO BE RELEVANT	
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP,A,O 019 600 (OY PARTEK AB) 26 November 1980 see page 1, line 1 - page 4, line 34	1-6
X	EP,A,O 399 320 (BAYER AG) 28 November 1990 see claims 1-3	1-6
X	EP,A,O 412 878 (ISOVER SAINT-GOBAIN) 13 February 1991 see page 2, line 1 - page 5, line 1	1-6
	-/	
:		
	-	

Further documents are listed in the continuation of box C.	X Patent family members are listed in annex.
'A' document defining the general state of the art which is not considered to be of particular relevance 'E' earlier document but published on or after the international filing date 'L' document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) 'O' document referring to an oral disclosure, use, exhibition or other means 'P' document published prior to the international filing date but	To later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention. "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone. "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. "&" document member of the same patent family
Date of the actual completion of the international search 3 October 1995	Date of mailing of the international search report 26. 10. 95
Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Van Bommel, L

Form PCT/ISA/210 (second sheet) (July 1992)

1

INTERNATIONAL SEARCH REPORT

Inter mal Application No
PCT/EP 95/02375

	·	PC1/Lr	95/02375
	auon) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages		Relevant to claim No.
A	GLASTECHNISCHE BERICHTE, vol. 64, no. 1, January 1991 FRANKFURT DE, pages 16-28, XP 000178832 R. M. POTTER ET AL 'Glass Fiber Dissolution in a Physiological Saline Solution' see page 26, left column, paragraph 3 - page 27, left column; table 2		Relevant to claim No.

INTERNATIONAL SEARCH REPORT

Information on patent family members

Inter. Jual Application No PCT/EP 95/02375

?

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-19600	26-11-80	SE-B- 418961 JP-C- 1195154 JP-A- 56014450 JP-B- 58024385 SE-A- 7904044 US-A- 4312952 US-A- 4381347	12-02-81 20-05-83 10-11-80 26-01-82
EP-A-399320	28-11-90	DE-A- 3917045 CA-A- 2017344 JP-A- 3005344 SU-A- 1813077 US-A- 5332698	25-11-90 11-01-91 30-04-93
EP-A-412878	13-02-91	FR-A- 2650821 FR-A- 2658182 AU-B- 630484 AU-B- 6002590 CA-A- 2022446 CN-A,B 1049834 CN-A- 1093066 DE-D- 69007369 DE-T- 69007369 ES-T- 2053139 HU-B- 210633 JP-A- 3093650 PL-B- 165859 SI-A- 9011548 US-A- 5108957	29-10-92 14-02-91 12-02-91 13-03-91 05-10-94 21-04-94 13-10-94 16-07-94 28-06-95 18-04-91 28-02-95 31-12-94 28-04-92